

## Folding of an Ala-Ala-Ala Tripeptide into a $\beta$ -Turn via Hydrophobic Encapsulation

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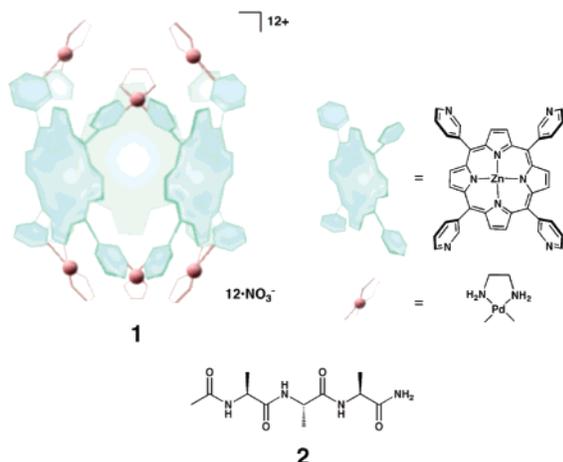
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$\beta$ -Turn conformation is one of the most important secondary structures of peptides. This fundamental motif is, like other peptide secondary structures, very unstable if a peptide exists as a short fragment.<sup>1</sup> The turn structure can be, in principle, constituted by four amino acid residues via hydrogen bonding between  $CO(i)$  and  $NH(i+3)$ . In general,  $\beta$ -turn conformation is stable in water only if it is involved in  $\beta$ -hairpin structures consisting of more than nine amino acid residues.<sup>2</sup> Here, we observe that even a three-residue peptide fragment, Ac-Ala-Ala-Ala-NH<sub>2</sub>, can be folded into a  $\beta$ -turn<sup>3</sup> (which is also regarded as a minimal  $3_{10}$ -helix) through encapsulation by a porphyrin-assembled synthetic host. Despite the presence of only one intramolecular hydrogen bond, the turn conformation is shown to be very stable because of efficient host–guest interaction.

For the efficient accommodation of a short peptide fragment, we employed prism-like porphyrin cage **1**<sup>4</sup> that provides a large hydrophobic binding pocket. This cage can be prepared by the self-assembly of tetrakis(3-pyridyl)-substituted porphyrin and (en)Pd-(NO<sub>3</sub>)<sub>2</sub>. The target tripeptide, Ac-Ala-Ala-Ala-NH<sub>2</sub> (**2**), contains only Ala that is the simplest  $\alpha$ -substituted amino acid residue.

The complex **1**·**2** was easily prepared by mixing the aqueous solutions of both components ([**1**] = [**2**] = 2 mM) at room temperature for a few minutes. Encapsulation of **2** within the cavity of **1** was confirmed by <sup>1</sup>H NMR analysis. Proton signals of complexed **2** were fully assigned from TOCSY and NOESY experiments. All the signals of **2** were considerably shifted upfield. In particular, the H $\beta$  protons of Ala1, Ala2, and Ala3 were observed around -5.0 ppm (Figure 1). This result suggested that all residues of peptide **2** were deeply encapsulated by the cavity of **1**. The association constant was estimated to be  $1 \times 10^6 \text{ M}^{-1}$  by <sup>1</sup>H NMR competition experiment.<sup>5</sup>



The turn conformation of **2** was elucidated from a NOESY experiment. NOE cross-peaks in NOESY were clearly observed

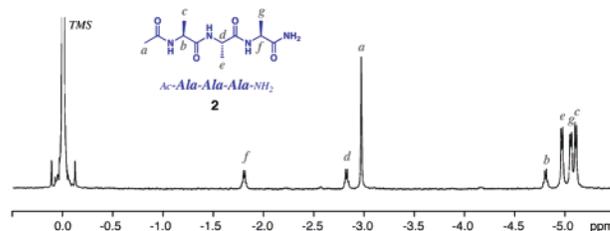


Figure 1. <sup>1</sup>H NMR spectrum of complex **1**·**2** in D<sub>2</sub>O (500 MHz, 2 mM, 27 °C, TMS as external standard).

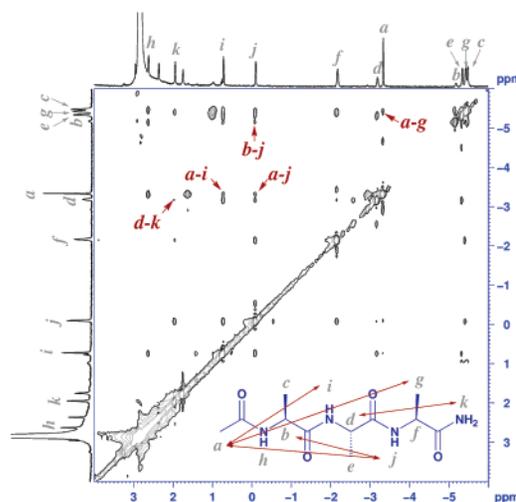
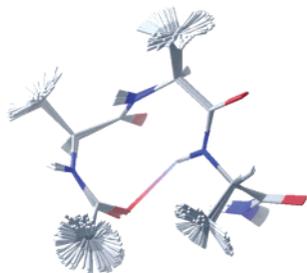


Figure 2. Selected NOESY spectrum (500 MHz, H<sub>2</sub>O/D<sub>2</sub>O = 9/1, 27 °C) of **1** ([**1**] = 2 mM) and **2** ([**2**] = 2 mM). Interresidue NOEs are denoted as letters in the spectrum and as arrows in the chemical structure of **2**.

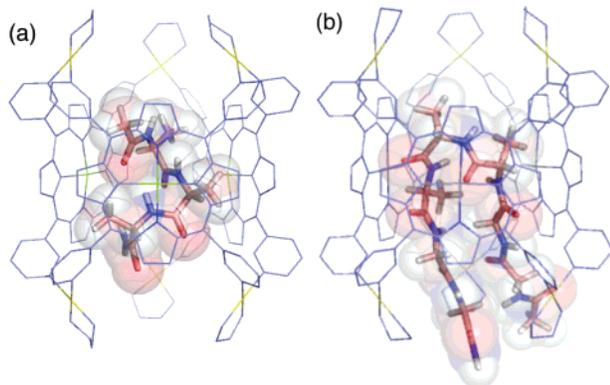
between the N-terminal acetyl group and Ala3 protons (HN and H $\beta$ ), indicating the proximity of the N- and C-terminal of **2** (Figure 2). The turn structure of **2** was also supported by <sup>3</sup>J<sub>NHCH $\alpha$  coupling constants. All <sup>3</sup>J<sub>NHCH $\alpha$  values were below 6 Hz, characteristic of helical or turn rather than random-coil conformation.<sup>6</sup></sub></sub>

Furthermore, the turn structure was strongly suggested by molecular dynamics (MD) simulation with the CNS program<sup>7</sup> that was run under restraints of NOE distances (11 intrasidue, 7 sequential, 5 medium range) and 3 backbone  $\phi$  angles from <sup>3</sup>J<sub>NHCH $\alpha$  values. Among randomly generated 50 initial structures, 48 of them were converged, with low backbone pairwise RMSD (0.09 Å), into almost identical lowest-energy structures that clearly showed the turn conformation (Figure 3). The MD calculation also suggested a hydrogen bond between the N-terminal acetyl group and NH of Ala3. This turn structure is a minimal  $3_{10}$ -helix and is also categorized as a  $\beta$ -turn conformation.<sup>8</sup></sub>

We confirmed that the MD-minimized structure fits the cavity of **1**. The crystal structure of **1**<sup>4</sup> and the MD-optimized structure of **2** were combined so that **2** was fully accommodated in the cavity



**Figure 3.** Superposition of the 48 lowest-energy structures by CNS<sup>7</sup> for bound peptide **2**.



**Figure 4.** Refined structures of (a) complex **1-2** and (b) complex **1-9** obtained after combining the crystal structure of **1** and the MD-minimized structures of **2** and **9**, respectively.<sup>9</sup>

of **1**. When the combined structure was refined,<sup>9</sup> the conformation of **2** remained almost unchanged, suggesting that host **1** provides an ideal cavity for recognizing the most stable conformation of **2** (Figure 4a). For comparison, tripeptide **2** in an extended conformation was also refined in the cavity of **1**, but efficient host–guest interactions were hardly observed.

In the refined **1-2** structure, the methyl groups of Ala residues and the porphyrin ligands showed efficient CH– $\pi$  contact, which presumably induces the folding of **2** into the  $\beta$ -turn conformation. In particular, the CH– $\pi$  contact of Ala2 seems to be the most important because the association constant of **1** with a singly Gly-mutated tripeptide, Ac-Ala-Gly-Ala-NH<sub>2</sub> (**3**), was considerably reduced ( $9 \times 10^3 \text{ M}^{-1}$ ). Other Gly-mutated tripeptides, Ac-Gly-Ala-Ala-NH<sub>2</sub> (**4**) and Ac-Ala-Ala-Gly-NH<sub>2</sub> (**5**), did not show significant decrease in association constants ( $8 \times 10^5$  and  $1 \times 10^6 \text{ M}^{-1}$  for **4** and **5**, respectively).<sup>5</sup> Binding was no longer observed for Ac-Gly-Gly-Gly-NH<sub>2</sub> (**6**) because of the absence of the CH– $\pi$  contact. Tripeptides, Ac-Val-Val-Val-NH<sub>2</sub> (**7**) and Ac-Leu-Leu-Leu-NH<sub>2</sub> (**8**), were also not bound at all because they are too bulky to be fit in the cavity of **1**.

Particularly interesting is that, through host–guest interaction with **1**, a stable  $\beta$ -hairpin-like structure is generated from a longer oligopeptide involving Ala-Ala-Ala sequence at the middle. When a heptapeptide, Ac-Gly-Gly-Ala-Ala-Ala-Gly-Gly-NH<sub>2</sub> (**9**), was treated with a molar equivalent of **1** in water, significant upfield shifts of proton signals were observed only around the Ala-Ala-Ala region ( $\Delta\delta = -6.2$ ,  $-6.5$ , and  $-5.5$  ppm for the methyl groups of Ala3, Ala4, and Ala5, respectively). The methylene groups of adjacent Gly2 and Gly6 were also shifted upfield ( $\Delta\delta = -6.2$  and

$-6.9$  ppm for Gly2;  $-4.2$  and  $-5.1$  ppm for Gly6), while terminal Ac-Gly1 and Gly7-NH<sub>2</sub> protons were hardly shielded. The association constant was  $8 \times 10^4 \text{ M}^{-1}$ .<sup>5</sup> MD simulation under NOE distance restraints followed by force-field optimization<sup>9</sup> clearly indicated a  $\beta$ -hairpin-like structure with the turn conformation at the Ala-Ala-Ala region, which should be deeply accommodated in the cavity of **1** (Figure 4b).<sup>10</sup>

In summary, self-assembled porphyrin cage **1** showed remarkable ability to fold Ala-Ala-Ala sequence into  $\beta$ -turn via encapsulation. The result demonstrated that peptide recognition within the large cavity of self-assembled cages<sup>11,12</sup> is a powerful method to produce the secondary structures of peptides even if the peptide fragment is considerably short. Folding relatively long peptide fragments into protein partial structures showing a protein's native activities is the ultimate goal of the present study.

**Supporting Information Available:** Experimental procedures, physical properties, NMR assignments, MD calculations, and binding studies (PDF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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